<u>Hazardous Waste Characteristics Tests</u>

<u>N/A</u>	Staff Yes No
1.1	Laboratory supervisor(s) with overall responsibility for sarea.
Na	me(s):
1.2	Analysts with Years of Experience in Parentheses
2.0	<u>Ignitability</u> <u>Equipment</u>
Date	<u>Manufacturer</u> <u>Model</u> <u>Installed</u>
<u>Du o c</u>	
2.1	Are manufacturers' operating manuals readily available?
2.2	Is there a laboratory reference standard?
2.3	What is the reference standard?
2.4	Are results of calibration kept in permanent bound logbook?
2.5	Has the instrument been modified in any way?
2.6	Are operations done in a vented area?
2.7	Is the test area free of drafts?
2.8	Is the sample container kept tightly sealed until the test portion is removed for analysis?
2.9	Are samples that flash after sustained heat from the flammability potential screen analysis (ASTM D4982) analyzed for a closed-cup flashpoint on a Pensky-Martens
	or Seta Flash closed-cup flash tester?
2.10	Does the Pensky-Martens apparatus meet specifications in ASTM D93, Annex Al on Apparatus Specifications?
2.11	Does the Seta Flash apparatus meet specifications in ASTM D3828, Annex A1 on Flash Test Apparatus?
3.0	Procedures & Practices for Flammability Potential Screen

Exposure to Heat and Flame (Screening Procedure)(ASTM D4982, Method A)

3.1	Is a means of extinguishing the flame readily available (e.g., soil or sand, large beaker?
3.2	Is the procedure performed with a bunsen burner or a propane torch?
3.3	Is an approximately 5 gram test portion used?
3.4	Is the flame held immediately above (not in contact with) the test portion for 2 to 3 seconds?
3.5	Then, is the flame applied to the test portion for at least 15 seconds?
3.6	Is ASTM D4982 documented in the lab's methods manual?
3.7	Is this procedure performed exactly as documented in Method A of ASTM D4982?

Yes No N/A Exposure to Spark Source (Screening Procedure) (ASTM D4982, Method B) 3.8 Is an approximately 100-mL test portion placed in a plastic beaker inside a steel vessel? 3.9 Is the test portion covered with a watchglass and allowed to stand at ambient conditions for 5 minutes? 3.10 Is the ambient temperature recorded? 3.11 Is the lid for the steeel vessel available for smothering flames for a positive test? 3.12 Is ASTM D4982 documented in the lab's methods manual? 3.13 Is this procedure performed exactly as documented in Method B of ASTM D4982? 3.14 Is a closed-cup flashpoint run on all samples that flash after sustained heating during the flammability potential screen? 3.0 Procedures & Practices Pensky-Martens Closed-cup Method (per ASTM D93 & SW-846 Method 1010) 3.1 Is the tester manual () or automatic ()? 3.2 In cases of dispute, is manual determination the referee method? 3.3 Are the 3 thermometers used according to specifications (ASTM E1 or IP thermometer specifications)? 3.4 Are samples stored in appropriate containers (not plastic)? 3.5 Is heating of specimens during preparation done with care to avoid loss? 3.6 Is dissolved or free water removed from the specimen? 3.7 Is meticulous attention paid to all details relating to the flame exposure device, size of test flame, rate of temperature increase, and rate of dipping the flame exposure device into the vapor of the specimen? 3.8 When the cup is in the stove is the locking device engaged? 3.9 Is the stirrer operating at 90 to 120 rpm? 3.10 Is the test flame inserted in the proper sequence?____ 3.11 Can the analyst demonstrate the "flash point" (does not confuse the true flash point with the bluish halo)?____ ____ 3.12 Is the stirrer operated at 250 ± 10 rpm for highly

viscous

materials?

- 3.13 Is the stirrer stopped when applying the test flame?
- 3.14 Is the ambient barometric pressure used for correction of flash point?
- 4.0 QA/QC for Pensky-Martens Closed-cup Method
 4.1 Are duplicates & std. reference mat'ls. routinely analyzed?
- 4.2 Is the flash point of p-xylene reference standard determined

in duplicate at least once per sample batch?

		<u>Yes No</u>
	Procedures & Practices for Setaflash Closed-Cup Meper (ASTM D3828 & SW-846 Method 1020) Is loss of volatile materials from sample avoided:	
3.2	Is the sample syringe 2.0 ± 0.1 mL?	
3.3	Is the use of plastic sample bottles avoided?	
3.4	Is heat transfer paste used?	
3.5	Is the test apparatus placed in subdued lighting a disturbing drafts?	and out of
3.6	Is the test apparatus standardized with a p-xylene n-butanol reference standard?	or
3.7	Is the barometric pressure correction made at the of the batch run?	beginning
3.8 safe	If sample cooling is required are the proper personety devices in use?	onal
3.9	Is a magnifying glass used to read thermometers?	
3.10 for	Is a step wise determination using a fresh sampl samples with a flash point at ambient or less?	e done
	<u>QA/QC for Seta Flash</u> When the flash/no flash method is used, does report	rt say so?
4.2	Are duplicates and reference standards routinely a	analyzed?
4.3	Is the flash point of the p-xylene or n-butanol restandard determined in duplicate at least once perbatch?	
11	Corrosivity	
_	Screening by pH Paper (ASTM D4980, Method A)	
3.0 3.1	<u>Procedures & Practices</u> For aqueous liquids, is the sample tested <u>without</u> of water?	addition
3.2	For solids, sludges, and non-aqueous liquids, is approximately 1 mL (or 1 g) of waste added to appr 10 mL of water and agitated for 10 seconds or untiwell mixed?	roximately
3.3	Is the sample placed onto the pH paper (i.e., not into the sample?	dipped
3.4	Has each batch of pH paper been verified by testing	na with

	pH buffer and is the result	documented?
3.5	Is ASTM D4980 documented in	the lab's methods manual?
3.6	Is this procedure performed Method A of ASTM D4980?	exactly as documented in

Yes No

N/A pH Screening by Electrometric Measurement (ASTM D4980, M	ethod
<u>B)</u>	
3.0 <u>Procedures & Practices</u> 3.1 Are the electrodes filled with sufficient electrolyte?	•
3.2 Are the electrodes stored in pH 7.0 buffer with KCl account as directed by manufacturer when not in use?	lded —— —
3.3 If the meter can determine slope, is the slope of the electrode documented?	
3.4 For aqueous liquids, is the sample tested <u>without</u> addiof water?	tion
3.5 For solids, sludges, and non-aqueous liquids, is the present made of the aqueous portion of the 10% slumwaste in water; e.g., the water layer from the water compatibility test (i.e., approximately 1 mL (or 1 g) waste added to approximately 10 mL of water and agitate for 10 seconds or until well mixed?	ry of of
3.6 Is ASTM D4980 documented in the lab's methods manual?_	
3.7 Is this procedure performed exactly as documented in Method B of ASTM D4980?	
(See also check sheets for Miscellaneous Electronic Probe	es)
Steel Coupons (SW 846 Method 1110A)	
 2.0 Equipment 2.1 Does the setup consist of a kettle or flask (500-5000 a reflux condenser, a thermowell and temperature regul device, a heating device and a coupon support system? 	
2.2 Are the test coupons physically and electrically insul from each other and from the setup?	lated
2.3 Is test coupon SAE 1020 steel of about 3.75 cm in diam	meter?
2.3 Is test coupon SAE 1020 steel of about 3.75 cm in diam 2.4 Is the test coupon exposed area known to ± 1 per cent?	
	•
2.4 Is the test coupon exposed area known to ± 1 per cent? 2.5 Is the test coupon cleaned prior to use by pickling	or or
 2.4 Is the test coupon exposed area known to ± 1 per cent? 2.5 Is the test coupon cleaned prior to use by pickling electrolytic removal or by grinding? 3.0 Procedures & Practices 3.1 Is the ratio of waste to surface at least 40 mL per cm 3.2 Is the test temperature 55°C? 3.3 After the 24 hour test period, is the surface cleaned all 	or or
 2.4 Is the test coupon exposed area known to ± 1 per cent? 2.5 Is the test coupon cleaned prior to use by pickling electrolytic removal or by grinding? 3.0 Procedures & Practices 3.1 Is the ratio of waste to surface at least 40 mL per cm 3.2 Is the test temperature 55°C? 3.3 After the 24 hour test period, is the surface cleaned 	or or

	elec	troly	tic	_?							
3.5	Is a	a blanl	c used	to	check	sound	metal	lost	during	cleaning?	
3.6	Are	dupli	cate s	ampl	es ana	alyzed	with	each a	analytic	cal batch?	

3.7 Is the proper formula used to calculate corrosion rate?_____

_ ___

N/A

Reactivity

Note. A solid waste exhibits the characteristic of reactivity if a representative sample has <u>any</u> of nine defined properties. For this program the tests for a cyanide or sulfide-bearing waste that may generate gases when exposed to pH conditions between 2 and 12.5 are the only portions of this characteristic that has interim guidance

Screening of Cyanides (ASTM D5049)

	<pre>Procedures & Practices If a sample is inorganic and has a pH greater than 6, is a cyanide screening analyis performed?</pre>
3.2	Are all positive screens followed up with further testing?
	
<u>Pr</u>	ussian Blue (ASTM D5049, Method B)
3.3	Are measured volumes (5 to 10 mL) of reagents added to the test portion?
3.4	Is the test portion made alkaline (pH 12-13) with 2.5N NaOH before any other readents are added?
3.5	Is ASTM D5049 documented in the lab's methods manual?
3.6	Is this procedure performed exactly as documented in Method B of ASTM D5049?
Су	antesmo Paper (ASTM D5049, Method C)
3.7	Is an approximately 5 g test portion placed in a flask or test tube and a water-moistened test paper strip suspended in the vessel?
3.8	Is the test portion acidified with 5 mL of 6N sulfuric acid and the vessel sealed with a stopper immediately after the test portion is acidified?
3.9	Is the test paper read after two minutes?
3.10	Then, is the pH of the acidified test portion verified to ensure that the pH is below 2.0?
3.11	. Is ASTM D5049 documented in the lab's methods manual?
3.12	Is this procedure performed exactly as documented in Method C of ASTM D5049?
<u>Ga</u>	s Detector Tubes (ASTM D5049, Method D)
3.13	Is approximately 50 mL of the buffer solution added to an approximately 20 g test portion?

2.3	Is pyridine-barbituric acid reagent less than 6 months old? 2.3.1 Is 2.3 documented?
	<u>Procedures & Practices</u> At time of collection are samples checked for oxidizing agents?
3.1.	1 If found, are they removed with ascorbic acid?
3.1.	2 Are samples preserved with 2 mL of 10N sodium hydroxide per liter of sample?
3.1.	3 Are samples refrigerated at 4°C?
3.2	Is the pretreatment for cyanides amenable to chlorination done in a hood?
3.3	Are the air bubbles in the distillation flask adjusted to two per second?

		<u>Yes</u>	No
<u>N/A</u> 3.4	Are steps taken to eliminate sulfides?		
	Are steps taken to eliminate nitrates and/or nitri	tes?	•
3.6	Is the color development time consistent at the consistency at the c	lor	
3.7	Are all standards distilled like the samples?		
3.8	If not, are at least a high and a low?		
3.9	Is sample distillation efficiency checked by spiki	ng?	
3.10	Are all standards distilled when developing curv samples with sulfide?	es f	or
	<pre>QA/QC Is one blank per sample batch done?</pre>		
4.2	Are check standards done after 15 samples?		
4.3	Is there one duplicate in every 10 samples?		
4.4	Are spiked samples used to check the procedure?		
4.5	Is the method of standard additions used when ther matrix interferences?	e ar	e
<u>Re</u>	active Sulfide (SW 846, Section 7.3.4 of Chapter 7)	
	<u>Procedures & Practices</u> Have samples been preserved with zinc acetate?		
3.2	Are precipitated samples done in the original bott	le?	
3.3	Are dewatered samples and the glass-fiber filter pin the original bottle?	aper	done
4.0 4.1	OA/OC Are calibration curves of one blank and three standone every hour of continuous analysis?	ıdard	ls
4.2	Is a blank run with each sample batch?		
4.3	Are check standards done every 15 samples?		
4.4	Is one duplicate done every ten samples?		
4.5	Are spiked samples used to check procedures?		

Free Liquids

Paint Filter Liquids Test (SW-846, Method 9095

3.0 Procedures & Practices

3.3 Is the test performed as mandated in SW-846, Method 9095?___

Yes No

N/A Toxic Characteristic Leaching Procedure (TCLP) SW-846, Method 1311)
2.0 Equipment 2.1 Agitation apparatus
2.1.1 Is the extraction vessel rotated end-over-end?
$2.\overline{1.2}$ Is the rotation at 30 \pm 2 rpm?
2.2 Extraction vessel (TCLP)
2.2.1 Is the vessel borosilicate glass?
2.2.2 Is the volume 2 L?
2.2.3 Are plastic bottles in use?
2.3 Extraction vessel (ZHE)
2.3.1 Is the zero-headspace extraction vessel (ZHE) used for volatile constituents?
2.3.2 Are the O-rings replaced frequently?
2.3.3 Is 2.3.4 documented?
2.3.4 Is the ZHE vessel pressure checked after each use?
2.3.5 Are psi to torque-inch-pounds conversion tables available?
2.4 Filtration 2.4.1 Are all filtrations done in a hood?
2.4.2 Are filtration devices fabricated from inert materials?
2.4.3 When evaluating wastes for volatiles is the ZHE used?
2.4.4 If the internal glass fiber filter ruptures, is an in-line glass fiber filter used?
2.4.5 Is the recommended filter device of 1.5L and equipped
to accommodate a 142 mm diameter filter in use?
2.4.6 Is positive pressure filtration practiced?
2.4.7 Are filters of borosilicate glass and without binders?
2.4.8 Are acid washed filters used with metal mobility valuations?

2.5 Extract collection devices
2.5.1 Is a TEDLAR bag used when the extract is aqueous?
2.5.2 — Is use of syringe avoided in 2.5.1 above?
2.5.3 If significant amounts of non-aqueous liquid phase is present the analyst may use the bag or a syringe. Does the analyst consistently use one device throughout the extraction?
2.6 Extraction fluids
2.6.2 Is fluid #2 documented?
3.0 Procedures & Practices for TCLP 3.1 Samples
3.1.1 Are 3 separate samples available for each determination?
3.1.2 Are samples free of added preservatives?
3.1.3 Is care taken to avoid loss of volatiles?
3.2 Preliminary evaluation
3.2.1 Is percent solids determined?
3.2.2 Is a determination made as to whether particle-size reduction is necessary?

3.4.12 Is extract aliquoted & stored with minimal headspace at 4°C?
4.0 <u>QA/QC</u>
4.1 Is a memory effect check done? 4.2 Are matrix spikes done to monitor the adequacy of methods?
4.3 Are QC procedures in use?
4.4 Method of Standard Additions 4.4.1 Is the method used if recovery is not between 50-150%?
4.4.2 Is the method used if the concentration measured in the extract is within 20% of the regulatory threshold?
4.4.3 Is a schedule available showing when a specific sample must be extracted and when the extract must be analyzed?

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